

Le 16/06/2017

Introduction to High Vacuum in Accelerators

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16/06/2017 B. Mercier – LAL Vacuum in Accelerators

Ecole Doctorale Pheniics

Introduction to High Vacuum in accelerators.

Part1: vacuum basis

- 1-1 Kinetic behavior of gas molecules
- 1-2 Mean free path / residence time
- 1-3 outgassing/desorption

Part3: Technologie evolution

- 3-1 Pressure distribution in accelerators
- 3-2 Distributed pumping
 - 3-2-a getter pump
 - 3-2-b NEG coating

Part2: vacuum in accelerators

- 2-1 interaction residual gas beam
 - 2-1-a Coulomb scattering
 - 2-1-b Bremsstrahlung
 - 2-1-C Ionization energy loss
 - 2-1-d ion accumulation
- 2-2 Interaction particules surface
 - 2-2-a Synchrotron radiation an Photon stimulated desorption
 - 2-2-b Ion stimulated Desorption
 - 2-2-c Electron stimulated Desorption an electron-cloud

What does vacuum mean?

Vacuum: an absence of matter ?

 Space where the molecules are strongly rarefied compared to atmospheric pressure

 10^{-9} mbar ~ 25 millions/cm³ at 20°C

Pollution source, background noise, parasitic interactions

quantum vacuum:

State minimum energy, vacuum fluctuation, ...

a vacuum: full of matter !!

Introduction :

The scale

Pressure $100 \text{ Pa}(\text{N/m}^2) = 1 \text{ mbar} = 0,75 \text{ Torr}$



Maxwell-Boltzmann velocity distribution





Monolayer formation time

And considering the molecule as spherical, for perfect atomic arrangement:

$$\tau = \frac{4}{\mathbf{d}^2 \cdot \mathbf{n} \cdot \mathbf{v}_{\mathbf{m}}}$$

(very approximate)

Order of magnitude one monolayer $\sim 10^{15}$ molécules/cm²

 $\tau \sim 40 \text{ min at P=10}^{-9} \text{ mbar } (N_2)$ $\tau \sim 3 \text{ sec at P=10}^{-6} \text{ mbar } (N_2)$



The concept of monolayer can appreciate the cleanliness of the surface

Mean Free Path

It is the path length that a molecules traverse between two succesives impacts with other molecules.

$$l_{\rm m} = \frac{1}{\sqrt{2}\pi.d^2.n}$$

$$_{\rm m} \sim 64$$
 km at P=10⁻⁹ mbar (N₂)
 $_{\rm m} \sim 6.5$ m at P=10⁻⁶ mbar (N₂)

summary Tables

Pour N₂ at 20°C

Pressure (hPa)	Strength per unit area	density molecular (molecule/cm ³)	Mean free path	Impingement rate (per cm ² and per sec)	Monolayer formation time
1013 1 10 ⁻³ 10 ⁻⁶	1 kg/cm ² 1 g/cm ² 1 mg/cm ² 1 mg/cm ²	2,50x10 ¹⁹ 2,47x10 ¹⁶ 2,47x10 ¹³ 2,47x10 ¹	0,1 μm 0,1 mm 10 cm	2,9x10 ²³ 2,9x10 ²⁰ 2,9x10 ¹⁷ 2,9x10 ¹⁴	3,4x10 ⁻⁹ s 3,4x10 ⁻⁶ s 3,4x10 ⁻³ s 3,4 s
10 ⁻⁹ 10 ⁻¹² 10 ⁻¹⁶ 10 ⁻²³	10 μg/m² 0,01 μg/m² 1 μg/km² 10 ⁻⁷ μg/km²	2,47x10 ⁷ 2,47x10 ⁴ 2,47 2,47 2,47	100 km 10⁵ km 10⁰ km 10¹6 km	2,9x10 ¹¹ 2,9x10 ¹¹ 2,9x10 ⁸ 2,9x10 ⁴ 2,9x10 ⁻³	1 heure 40 jours 1000 ans 10 ¹⁰ ans
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outgassing

Outgassing is the spontaneous evolution of gas from solid or liquid.



desorption /degazing

<u>Desorption</u> is the release of adsorbed chemical species from the surface of a solid or liquid.



Vacuum basis :

Desorption: gas flow

At the thermodynamic equilibrium

Gas flow

$$Q = P \frac{dV}{dt} = PS = \frac{dN}{dt} KT$$

S effectif pumping speed N molecule number

 $Pa.m^{3}/s$ (=10 mbar.l/s)

The outgassing rate τ : a gas flow per unit of surface

 $\tau = Q / A$ A desorption area

$$Pa.m^{3}.s^{-1}.m^{-2} = Pa.m.s^{-1} (=10^{-3} mbar.l.s^{-1}.cm^{-2})$$

Importance of the gas flow

At the thermodynamic equilibrium, the pressure in the system is expressed by:

 $P = \frac{\sum Q}{S} + P_0$

Decrease of the work presssure P

☐ Increase the pumping speed

 P_0 limit pressure of pumping system

S effectif pumping speed

Approximately 2 to 3 orders of magnitude with a high cost and limited effectiveness conductances

About 10 orders of magnitude depending on the choice of materials and treatments

Attention to minority surfaces

First of all decrease the degassing rates

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Vacuum basis :

Time dependance of the outgassing flow rate for différent materials



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Vacuum basis :

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The pressure vs. Time behavior of a typical vacuum system (metal chamber)



Vacuum in accelerators :



Vacuum in accelerators :



Beam & Vacuum & Surface: a difficult coexistence!

Interaction: Beam - résidual gas



Interaction: Beam- residual gas



Interaction: Beam- residual gas

Elastic interaction – Single Coulomb scattering



Emittance is proportional to residual gas pressure and depends sqare residual gas atomic number



Nuclear scattering LHC

Life time limit ~100h P~10-8 mbar H_2 equivalent 80 mW/m heat load in the cold mass

Vacuum in accelerators :



A very strong dependence on the target atomic number is observed, which means that even a small amount of a heavy gas may have a higher influence on the lifetime than the usually dominating hydrogen content.

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Loss beam energy

Secondary particule emission: electrons, ions

→ Interaction on the surface and desorption

Vacuum in accelerators :

Ion accumulation

negative charge beam

An important phenomenon for negative charges beam, for high current and for short bunch spacing

Pressure due to the ion containment



Ion accumulation

<u>Ions stability criterion :</u>

Calculation of critical mass ¹:

[1] Neutralisation of accelerator beams by ionisation of the residual gas Y. Baconnier, A. Poncet and P.F. Tavares CERN

All ions of mass greater than the critical mass will be accumulated.

 $A_{c} = \frac{N_{e}}{N_{h}^{2}} \cdot r_{p} \cdot \frac{2 \cdot \pi \cdot R}{\beta \cdot b^{2} \cdot (1 + a/b)}$ avec Ne 2.π.R ring circum

avec Ne e- number, N_b bunch number, r_p proton radius, 2. π .R ring circumference, β ratio of v to the speed of light c, a and b beam size.

Neutralizing Factor

η= <u>number of positive charge (ions)</u> Number of negative charge (beam electron)



Vacuum in accelerators :

Accélératrice

Section

Canon_e-

Ion accumulation

Ionic pressure seen by beam

Increased pressure due to the ion density.

$$P_{ion} = \frac{I}{e} \cdot \frac{\eta^* \cdot k \cdot T}{\beta \cdot C \cdot \pi \cdot 2 \cdot \sigma_x \cdot \sigma_y}$$

Ionic pressure (mbar)





Electrodes + gap between two injections (5μ s)

C. Bruni, J. Haissinski, T. Demma

Vacuum in accelerators :

Synchrotron radiation

Any charged particle undergoing acceleration, centripetal or longitudinal, produces radiation.

> **RAYONNEMENT SYNCHROTRON :** DIPOLES

Main source on circular colliders



Emission vers l'avant dans un angle très faible $\alpha \frac{1}{E}$

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Spectre blanc

Power loss $P_0 [W/m] = 88.57 \frac{E[GeV]^4}{2\pi \rho[m]^2} I[mA]$ electron $P_0 [W/m] = 7.79 \ 10^{-12} \ \frac{E[GeV]^4}{2\pi \ \rho[m]^2} I[mA]$ proton

Critical energy

Electrons :
$$\varepsilon_c[eV] = 2.218 \ 10^3 \frac{E[GeV]^3}{\rho[m]}$$

Protons : $\varepsilon_c[eV] = 3.5835 \ 10^{-7} \frac{E[GeV]^3}{\rho[m]}$
• $\Gamma[photons.m^{-1}.s^{-1}] = 1.28810^{17} \frac{E[GeV]}{\rho[m]} I[mA]$
• $\Gamma[photons.m^{-1}.s^{-1}] = 7.01710^{13} \frac{E[GeV]}{\rho[m]} I[mA]$

synchrotron radiation

some examples

		Soleil	KEK-B		LEP		LHC		
			LER	HER	Inj.	1	2	Inj.	Col.
Particule		e⁻	e ⁺	e⁻	e⁻	e-	e	р	р
Courant	mA	500	2600	1100	3	3	7	584	584
Energie	GeV	2.75	3.5	8	20	50	96	450	7000
R. courbure	m	5.36	16.31	104.46		2962.96		2784	.302
Puissance	W/m	14 030	20 675	5 820	0.8	30	955	0	0.2
En. critique	eV	8 600	5 800	11 000	6 000	94 000	660 000	0	44
Flux	photons/m/s	3 10 ¹⁹	7 10 ¹⁹	1 10 ¹⁹	3 10 ¹⁵	7 10 ¹⁵	3 1016	7 10 ¹⁵	1 1017
Dose a 3000 h	photons/m	4 10 ²⁶	8 10 ²⁶	1 10 ²⁶	3 10 ²²	7 10 ²²	3 10 ²³	7 10 ²²	1 10 ²⁴

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Photon stimulated desorption

photodesorption Yield $\boldsymbol{\eta}$

 $\eta = \frac{\text{desorbed molecule}}{\text{Incident photons}}$

photodesorption Yield η at machine start-up



material type, surface finish, energy and particle type, incidence angle, particule dose

Experimental determination

the pressure distribution in Cell HER (SUPERB) with synchrotron radiation at machine start-up photodesorption

Conditioning - Scrubbing

Surface conditioning according to the received dose



Vacuum in accelerators :

Photon stimulated desorption

Important implications for the design







Figure 1: Typical structure of ante-chamber for the LER of Super-KEKB.

SOLEIL

Puissance	W/m	14 030	
En. critique	eV	8 600	
Flux	photons/m/s	3 10 ¹⁹	

Absorbing power "crotch" : GLIDCOP copper cooled by water(256 W / mm2)

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Photon stimulated desorption

LHC Dipole Vacuum System

- Cold bore (CB) at 1.9 K which ensures leak tightness
- Beam screen (BS) at 5-20 K which intercepts thermal loads and acts as

a screen

LHC DIPOLE : STANDARD CROSS-SECTION





Courtesy N. Kos CERN AT/VAC

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Vacuum in accelerators :

interaction particules- surface



Electron-Cloud / Electron Stimulated Desorption

An important phenomenon for positive charges beam, for high current and for short bunch spacing

Main parameters:



Geometry chamber (antichamber)

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Y. Suetsugu SuperKEKB SuperKEKB Dipole Chamber Extrusion

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Electron-Cloud / Electron Stimulated Desorption

LHC





Electric field Interaction - surface

Nonzero Impédance of the chamber



Accentuate by section changes, discontinuity ..



size evolution of vacuum chambers on Accelerators

Size vacuum chambers are constrained by the beam of performance expected by the magnetic system including:

- Strong focus therefore reduces throat circle
- Low distance between the magnets.



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Consequences : size room more and more small Significant decrease in conductance and increasing the maximum pressure





Non evaporable Getter (NEG)

Principle

coating of metal alloy (NEG) on the walls of vacuum chambers or on a strip

• coating Zr-Al, Zr-V-Fe, Ti-Zr-V,



distributed pumping

Evolution vacuum technology

NEG strips

St101 NON-EVAPORABLE GETTER (Zr Al)

activation temperature at 700°c

St707 NON-EVAPORABLE GETTER (Zr 70-V

24.6-Fe 5.4 wt%) activation temperature at 450°c

NEG materials are embedded in a constantan ribbon that is heated by the Joule effect



P< 10⁻¹¹ mbar

The high temperature activation NEG can generate mechanical stresses in the vacuum chamber.

Need to have an antechamber

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NEG strips

Design and construction of the SuperKEKB vacuum system

Yusuke Suetsugu,^{a)} Ken-ichi Kanazawa, Kyo Shibata, Takuya Ishibashi, Hiromi Hisamatsu, Mitsuru Shirai, and Shinji Terui *High Energy Accelerator Research Organization (KEK), Tsukuba, Ibaraki 305-0801, Japan*



FIG. 4. (Color online) Multilayered NEG strips for the antechamber.



Fig. 1. Example of a beam pipe with antechambers and distributed pumps.

St707 NON-EVAPORABLE GETTER

J. Vac. Sci. Technol. A, Vol. 30, No. 3, May/Jun 2012

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NEG coating in vacuum chamber

- Titane Zirconium Vanadium coating
- Low activation temperature Ta=180°C

surface of vacuum chamber becomes a pump!!!



- SOLEIL: NEG 60 % of the circumference
- LHC: 6 km in warm part

MAX IV: 95% of the circumference



Coating at LAL (~2 µm)

distributed pumping

NEG coating-vacuum performances

Titane ZirconiumVanadium coating

Electron and photon stimulated desorption photons (ESD and PSD) are reduced A low secondary electron yield (SEY)



P. Chiggiato, R. Kersevan / Vacuum 60 (2001) 67-72

distributed pumping

NEG coating-vacuum disadvantage

Titane ZirconiumVanadium coating



(No pumping of rare gas, CH₄)

saturation effect



P. Chiggiato, R. Kersevan / Vacuum 60 (2001) 67-72

Conclusion

Trend: High luminosity accelerators (involves small-sized beams, high current and a high repetition frequency).

Many beams interactions, residual gas and surface will generate induced outgassing sources and impact the shape and the beam energy

Reduction of induced outgassing sources

Geometry, coating, surface condition, scrubbing,.....

Increase the effective pumping speed

Distributed pumping NEG strips Neg coating (TiZrV)

Les devises Shadok



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- J. Wiley & sons. Elsevier Science.
- Vacuum Technology, A. Roth. Elsevier Science.
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•L'ultra vide des grands instruments – V. Baglin Gif-sur-Yvette - 22 septembre2009

Journals

- Journal of vacuum science and technology
- Vacuum
- Physical review special topics Accelerators and Beam



Le dépôt NEG

La pulvérisation cathodique magnétron







Paramètres de dépôt B = 100 G

V = -500 V

Pression Krypton : 2 10⁻² mbar

Vitesse de dépôt : 1 Å/s

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Electron Cloud Effects









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POMPES IONIQUES

Trois phénomènes de pompage :

- par diffusion (principalement H2)

Principalement deux configurations:

- Pompes ioniques Diodes (et noble diode)
- Pompes ioniques triodes (et starcell)





lons à très haute énergie



• La désorption est déterminée par l'énergie donnée aux électrons (pouvoir d'arrêt électronique)



• La désorption induite par les électrons est le mécanisme responsable

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Mécanisme

• Effet de surface (sauf diffusion H₂) dû à une activation thermique

• « Inelastic thermal spike model » : carte de température couplée au modèle de désorption thermique

$$\eta_{calculated} = 185$$

Remèdes

- Utilisation de revêtements NEG (LEIR, RHIC, GSI)
- Intercepter les ions « perdus » sur des collimateurs dédies :
 - LEIR : plaquage d'or de 30 μm sur de l'acier 316 LN, incidence perpendiculaire
 - GSI : plaquage d'or de 0.1 μm, incidence perpendiculaire. Absorbeur introduit dans une chambre secondaire. Revêtement NEG



Utiliser le conditionnement par le faisceau

Figure 2: Horizontal cut through the installed SIS18 ion catcher prototype. Yellow: beam, red: secondary chamber, brown: beam absorbers.

C. Omet et. al. EPAC 2008, Genoa, Italy

Vapor pressure

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Energie critique 'collision élastique/ émission radiative) en fonction de Z



Perte d'énergie des électrons dans du plomb par unité de radiation X0. L'énergie critique est de 7,43 MeV.



Plomb 11,35 g/cm3 (20°C)

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